

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
12 September 2002 (12.09.2002)

PCT

(10) International Publication Number
WO 02/070630 A1

(51) International Patent Classification⁷: **C10G 65/12, 65/04**

(21) International Application Number: **PCT/EP02/02449**

(22) International Filing Date: **5 March 2002 (05.03.2002)**

(25) Filing Language: **English**

(26) Publication Language: **English**

(30) Priority Data:
01400562.3 5 March 2001 (05.03.2001) EP
01402181.0 16 August 2001 (16.08.2001) EP

(71) Applicant (*for all designated States except US*): **SHELL INTERNATIONALE RESEARCH MAATSCHAPPIJ B.V.** [NL/NL]; Carel van Bylandtlaan 30, NL-2596 HR The Hague (NL).

(72) Inventors; and

(75) Inventors/Applicants (*for US only*): **GERMAINE, Gilbert, Robert, Bernard** [FR/FR]; Route de departementale n° 3, F-76650 Petit Couronne (FR). **WEDLOCK, David, John** [GB/GB]; Pool Lane, Ince, Chester, Cheshire CH2 4NU (GB).

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZM, ZW.

(84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

- *with international search report*
- *before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments*

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: **PROCESS TO PREPARE A WAXY RAFFINATE**

(57) Abstract: Process to prepare a waxy raffinate product by (a) hydrocracking/hydroisomerising a Fischer-Tropsch derived feed, wherein weight ratio of compounds having a least 60 or more carbon atoms and compounds having at least 30 carbon atoms in the Fischer-Tropsch product is at least 0.2 and wherein at least 30 wt% of compounds in the Fischer-Tropsch derived feed have at least 30 carbon atoms, (b) isolating from the product of step (a) a waxy raffinate product having a T10 wt% boiling point of between 200 and 450°C and a T90 wt% boiling point of between 400 and 650 °C.

WO 02/070630 A1

PROCESS TO PREPARE A WAXY RAFFINATE

The invention is directed to a process to prepare a waxy raffinate from a Fischer-Tropsch product. The waxy raffinate product as obtained in this process may find application as a feedstock to prepare a lubricating base oil. Said preparation of the base oil and the preparation of the waxy raffinate product may take place at different locations. Suitably the waxy raffinate product is prepared at the location where the Fischer-Tropsch product is prepared and the lubricating base oil is prepared at a location near the main markets for these products. Generally these locations will be different resulting in that the waxy raffinate products will have to be transported, for example by ship, to the lubricant base oil manufacturing location. This manner of preparing base oils is advantageous because only one product has to be shipped to the potential base oil and lubricant markets instead of transporting the various base oils grades which may be prepared from the waxy raffinate product. Applicants have now found a process to prepare such a waxy raffinate product, which is transportable and from which a novel class of base oils can be prepared.

Prior art base oils as described in for example WO-A-0014179, WO-A-0014183, WO-A-0014187 and WO-A-0014188 comprise at least 95 wt% of non-cyclic isoparaffins. WO-A-0118156 describes a base oil derived from a Fischer-Tropsch product having a naphthenics content of less than 10%. Also the base oils as disclosed in applicant's patent applications EP-A-776959 or EP-A-668342 have been found to comprise less than 10 wt% of cyclo-paraffins. Applicants repeated Example 2 and 3 of EP-A-776959 and base oils were obtained, from a waxy Fischer-Tropsch

synthesis product, wherein the base oils consisted of respectively about 96 wt% and 93 wt% of iso- and normal paraffins. Applicants further prepared a base oil having a pour point of -21 °C by catalytic dewaxing a Shell MDS Waxy Raffinate (as obtainable from Shell MDS Malaysia Sdn Bhd) using a catalyst comprising synthetic ferrierite and platinum according to the teaching of EP-A-668342 and found that the content of iso- and normal paraffins was about 94 wt%. Thus these prior art base oils derived from a Fischer-Tropsch synthesis product had at least a cyclo-paraffin content of below 10 wt%. Furthermore the base oils as disclosed by the examples of application WO-A-9920720 will not comprise a high cyclo-paraffin content. This because feedstock and preparation used in said examples is very similar to the feedstock and preparation to prepare the above prior art samples based on EP-A-776959 and EP-A-668342.

Applicants have now found a method to prepare a waxy raffinate product, from which lubricating base oil composition can be prepared having a higher cyclo-paraffin content and a resulting improved solvency when compared to the disclosed base oils. This is found to be advantageous in for example industrial formulations such as turbine oils and hydraulic oils comprising for the greater part the base oil according to the invention. Furthermore the base oil compositions will cause seals in for example motor engines to swell more than the prior art base oils. This is advantageous because due to said swelling less lubricant loss will be observed in certain applications. Applicants have found that such a base oil is an excellent API Group III base oil having improved solvency properties.

The invention is directed to the following process.
Process to prepare a waxy raffinate product by

(a) hydrocracking/hydroisomerising a Fischer-Tropsch derived feed, wherein weight ratio of compounds having at least 60 or more carbon atoms and compounds having at least 30 carbon atoms in the Fischer-Tropsch product is at least 0.2 and wherein at least 30 wt% of compounds in the Fischer-Tropsch derived feed have at least 30 carbon atoms,

(b) isolating from the product of step (a) a waxy raffinate product having a T10 wt% boiling point of between 200 and 450 °C and a T90 wt% boiling point of between 400 and 650 °C.

Applicants found that by performing the hydrocracking/hydroisomerisation step with the relatively heavy feedstock a waxy raffinate product is obtained from which valuable products may be prepared, such as the base oil product as described in this application. A further advantage is that both fuels, for example gas oil, and a waxy raffinate product suited for preparing base oils are prepared in one hydrocracking/hydroisomerisation process step.

The process of the present invention also results in middle distillates having exceptionally good cold flow properties. These excellent cold flow properties could perhaps be explained by the relatively high ratio iso/normal and especially the relatively high amount of di- and/or trimethyl compounds. Nevertheless, the cetane number of the diesel fraction is more than excellent at values far exceeding 60, often values of 70 or more are obtained. In addition, the sulphur content is extremely low, always less than 50 ppmw, usually less than 5 ppmw and in most case the sulphur content is zero. Further, the density of especially the diesel fraction is less than 800 kg/m³, in most cases a density is observed between 765 and 790 kg/m³, usually around 780 kg/m³ (the

viscosity at 100 °C for such a sample being about 3.0 cSt). Aromatic compounds are virtually absent, i.e. less than 50 ppmw, resulting in very low particulate emissions. The polyaromatic content is even much lower than the aromatic content, usually less than 1 ppmw. T95, in combination with the above properties, is below 380 °C, often below 350 °C.

The process as described above results in middle distillates having extremely good cold flow properties. For instance, the cloud point of any diesel fraction is usually below -18 °C, often even lower than -24 °C. The CFPP is usually below -20 °C, often -28 °C or lower. The pour point is usually below -18 °C, often below -24 °C.

The relatively heavy Fischer-Tropsch derived feed as used in step (a) has at least 30 wt%, preferably at least 50 wt%, and more preferably at least 55 wt% of compounds having at least 30 carbon atoms. Furthermore the weight ratio of compounds having at least 60 or more carbon atoms and compounds having at least 30 carbon atoms of the Fischer-Tropsch derived feed is at least 0.2, preferably at least 0.4 and more preferably at least 0.55. The Fischer-Tropsch derived feed is preferably derived from a Fischer-Tropsch product which comprises a C₂₀⁺ fraction having an ASF-alpha value (Anderson-Schulz-Flory chain growth factor) of at least 0.925, preferably at least 0.935, more preferably at least 0.945, even more preferably at least 0.955.

The initial boiling point of the Fischer-Tropsch derived feed may range up to 400 °C, but is preferably below 200 °C. Preferably at least any compounds having 4 or less carbon atoms and any compounds having a boiling point in that range are separated from a Fischer-Tropsch synthesis product before the Fischer-Tropsch synthesis product is used as a Fischer-Tropsch derived feed in

step (a). The Fischer-Tropsch derived feed as described in detail above will for the greater part comprise of a Fischer-Tropsch synthesis product, which has not been subjected to a hydroconversion step as defined according to the present invention. The content of non-branched compounds in the Fischer-Tropsch synthesis product will therefore be above 80 wt%. In addition to this Fischer-Tropsch product also other fractions may be part of the Fischer-Tropsch derived feed. Possible other fractions may suitably be any high boiling fraction obtained in step (b) or any surplus waxy raffinate product, which cannot be shipped away to lubricating manufactures. By recycling this fraction additional middle distillates may be prepared.

Such a Fischer-Tropsch derived feed is suitably obtained by a Fischer-Tropsch process, which yields a relatively heavy Fischer-Tropsch product. Not all Fischer-Tropsch processes yield such a heavy product. An example of a suitable Fischer-Tropsch process is described in WO-A-9934917 and in AU-A-698392. These processes may yield a Fischer-Tropsch product as described above.

The Fischer-Tropsch derived feed and the resulting waxy raffinate product will contain no or very little sulphur and nitrogen containing compounds. This is typical for a product derived from a Fischer-Tropsch reaction, which uses synthesis gas containing almost no impurities. Sulphur and nitrogen levels will generally be below the detection limits, which are currently 5 ppm for sulphur and 1 ppm for nitrogen.

The Fischer-Tropsch derived feed may optionally be subjected to a mild hydrotreatment step in order to remove any oxygenates and saturate any olefinic compounds present in the reaction product of the Fischer-Tropsch reaction. Such a hydrotreatment is described in

EP-B-668342. The mildness of the hydrotreating step is preferably expressed in that the degree of conversion in this step is less than 20 wt% and more preferably less than 10 wt%. The conversion is here defined as the weight percentage of the feed boiling above 370 °C, which reacts to a fraction boiling below 370 °C. After such a mild hydrotreatment lower boiling compounds, having four or less carbon atoms and other compounds boiling in that range, will preferably be removed from the effluent before it is used in step (a).

The hydrocracking/hydroisomerisation reaction of step (a) is preferably performed in the presence of hydrogen and a catalyst, which catalyst can be chosen from those known to one skilled in the art as being suitable for this reaction. Catalysts for use in step (a) typically comprise an acidic functionality and a hydrogenation/dehydrogenation functionality. Preferred acidic functionality's are refractory metal oxide carriers. Suitable carrier materials include silica, alumina, silica-alumina, zirconia, titania and mixtures thereof. Preferred carrier materials for inclusion in the catalyst for use in the process of this invention are silica, alumina and silica-alumina. A particularly preferred catalyst comprises platinum supported on a silica-alumina carrier. If desired, applying a halogen moiety, in particular fluorine, or a phosphorous moiety to the carrier, may enhance the acidity of the catalyst carrier. Examples of suitable hydrocracking/hydroisomerisation processes and suitable catalysts are described in WO-A-0014179, EP-A-532118, EP-A-666894 and the earlier referred to EP-A-776959.

Preferred hydrogenation/dehydrogenation functionalities are Group VIII non-noble metals, for example nickel and cobalt, optionally in combination with molybdenum or copper, and Group VIII noble metals, for

example palladium and more preferably platinum or platinum/palladium alloys. The catalyst may comprise the noble metal hydrogenation/dehydrogenation active component in an amount of from 0.005 to 5 parts by weight, preferably from 0.02 to 2 parts by weight, per 100 parts by weight of carrier material. A particularly preferred catalyst for use in the hydroconversion stage comprises platinum in an amount in the range of from 0.05 to 2 parts by weight, more preferably from 0.1 to 1 parts by weight, per 100 parts by weight of carrier material. The catalyst may also comprise a binder to enhance the strength of the catalyst. The binder can be non-acidic. Examples are clays and other binders known to one skilled in the art.

In step (a) the feed is contacted with hydrogen in the presence of the catalyst at elevated temperature and pressure. The temperatures typically will be in the range of from 175 to 380 °C, preferably higher than 250 °C and more preferably from 300 to 370 °C. The pressure will typically be in the range of from 10 to 250 bar and preferably between 20 and 80 bar. Hydrogen may be supplied at a gas hourly space velocity of from 100 to 10000 Nl/l/hr, preferably from 500 to 5000 Nl/l/hr. The hydrocarbon feed may be provided at a weight hourly space velocity of from 0.1 to 5 kg/l/hr, preferably higher than 0.5 kg/l/hr and more preferably lower than 2 kg/l/hr. The ratio of hydrogen to hydrocarbon feed may range from 100 to 5000 Nl/kg and is preferably from 250 to 2500 Nl/kg.

The conversion in step (a) as defined as the weight percentage of the feed boiling above 370 °C which reacts per pass to a fraction boiling below 370 °C, is at least 20 wt%, preferably at least 25 wt%, but preferably not more than 80 wt%, more preferably not more than 70 wt%. The feed as used above in the definition is the total hydrocarbon feed fed to step (a), thus also any optional

recycle of the higher boiling fraction as obtained in step (b).

5 In step (b) the product of step (a) is separated into one or more gas oil fractions, a waxy raffinate product having a T10 wt% boiling point of between 200 and 450 °C and a T90 wt% boiling point of between 400 and 650 °C and more preferably a T90wt% boiling point of below 550 °C. Depending on the conversion in step (a) and the properties of the total feed to step (a) also a higher
10 boiling fraction may be obtained in step (b).

The separation in step (b) is preferably performed by means of a first distillation at about atmospheric conditions, preferably at a pressure of between 1.2-2 bara, wherein the gas oil product and lower boiling
15 fractions, such as naphtha and kerosine fractions, are separated from the higher boiling fraction of the product of step (a). The higher boiling fraction, of which suitably at least 95 wt% boils above 370 °C, is subsequently further separated in a vacuum distillation
20 step wherein a vacuum gas oil fraction, the waxy raffinate product and the higher boiling fraction are obtained. The vacuum distillation is suitably performed at a pressure of between 0.001 and 0.05 bara.

The vacuum distillation of step (b) is preferably
25 operated such that the desired waxy raffinate product is obtained boiling in the specified range and having a kinematic viscosity at 100 °C of preferably between 3 and 10 cSt.

The waxy raffinate product as obtained by the above
30 process has properties, such as pour point and viscosity, which makes it suitable to be transported, suitable by ships, to a lubricating base oil manufacturing location. Preferably the waxy raffinate is stored and transported in the absence of oxygen such to avoid oxidation of the
35 paraffin molecules present in the waxy raffinate product.

Suitable nitrogen blanketing is applied during said storage and transport. Preferably the waxy raffinate product has a pour point of above 0 °C. This makes it possible to transport the waxy raffinate as a solid by
5 for example keeping the product at ambient temperatures. Transporting the product in the solid state is advantageous because it further limits the ingress of oxygen and thus avoids oxidation. Means to liquefy the product at the unloading facility should be present.
10 Preferably indirect heating means such as steam heated coils are present in the storage tanks, such that the product may be liquefied before being discharged from the tanks. Transport lines are also preferably provided with means to keep the product in a liquid state.

15 The waxy raffinate product may find various applications. A most suited application is to use the waxy raffinate product as feedstock to prepare lubricating base oils by subjecting the waxy raffinate product to a pour point reducing step. Optionally the
20 waxy raffinate product may be blended with slack wax in order to upgrade the slack wax properties with respect to sulphur, nitrogen and saturates content before subjecting the waxy raffinate to a pour point reducing step.

25 With a pour point reducing treatment is understood every process wherein the pour point of the base oil is reduced by more than 10 °C, preferably more than 20 °C, more preferably more than 25 °C.

The pour point reducing treatment can be performed by means of a so-called solvent dewaxing process or by means
30 of a catalytic dewaxing process. Solvent dewaxing is well known to those skilled in the art and involves admixture of one or more solvents and/or wax precipitating agents with the waxy raffinate product and cooling the mixture to a temperature in the range of from -10 °C to -40 °C,
35 preferably in the range of from -20 °C to -35 °C, to

- 10 -

separate the wax from the oil. The oil containing the wax is usually filtered through a filter cloth which can be made of textile fibres, such as cotton; porous metal cloth; or cloth made of synthetic materials. Examples of solvents which may be employed in the solvent dewaxing process are C₃-C₆ ketones (e.g. methyl ethyl ketone, methyl isobutyl ketone and mixtures thereof), C₆-C₁₀ aromatic hydrocarbons (e.g. toluene), mixtures of ketones and aromatics (e.g. methyl ethyl ketone and toluene), autorefrigerative solvents such as liquefied, normally gaseous C₂-C₄ hydrocarbons such as propane, propylene, butane, butylene and mixtures thereof. Mixtures of methyl ethyl ketone and toluene or methyl ethyl ketone and methyl isobutyl ketone are generally preferred. Examples of these and other suitable solvent dewaxing processes are described in Lubricant Base Oil and Wax Processing, Avilino Sequeira, Jr, Marcel Dekker Inc., New York, 1994, Chapter 7.

A preferred pour point reducing process is the catalytic dewaxing process. With such a process it has been found that base oils having a pour point of even below -40 °C can be prepared when starting from the waxy raffinate product according to the present process.

The catalytic dewaxing process can be performed by any process wherein in the presence of a catalyst and hydrogen the pour point of the waxy raffinate product is reduced as specified above. Suitable dewaxing catalysts are heterogeneous catalysts comprising a molecular sieve and optionally in combination with a metal having a hydrogenation function, such as the Group VIII metals. Molecular sieves, and more suitably intermediate pore size zeolites, have shown a good catalytic ability to reduce the pour point of the waxy raffinate product under catalytic dewaxing conditions. Preferably the

- 11 -

intermediate pore size zeolites have a pore diameter of between 0.35 and 0.8 nm. Suitable intermediate pore size zeolites are mordenite, ZSM-5, ZSM-12, ZSM-22, ZSM-23, SSZ-32, ZSM-35 and ZSM-48. Another preferred group of molecular sieves are the silica-aluminophosphate (SAPO) materials of which SAPO-11 is most preferred as for example described in US-A-4859311. ZSM-5 may optionally be used in its HZSM-5 form in the absence of any Group VIII metal. The other molecular sieves are preferably used in combination with an added Group VIII metal. Suitable Group VIII metals are nickel, cobalt, platinum and palladium. Examples of possible combinations are Pt/ZSM-35, Ni/ZSM-5, Pt/ZSM-23, Pd/ZSM-23, Pt/ZSM-48 and Pt/SAPO-11. Further details and examples of suitable molecular sieves and dewaxing conditions are for example described in WO-A-9718278, US-A-4343692, US-A-5053373, US-A-5252527 and US-A-4574043.

The dewaxing catalyst suitably also comprises a binder. The binder can be a synthetic or naturally occurring (inorganic) substance, for example clay, silica and/or metal oxides. Natural occurring clays are for example of the montmorillonite and kaolin families. The binder is preferably a porous binder material, for example a refractory oxide of which examples are: alumina, silica-alumina, silica-magnesia, silica-zirconia, silica-thoria, silica-beryllia, silica-titania as well as ternary compositions for example silica-alumina-thoria, silica-alumina-zirconia, silica-alumina-magnesia and silica-magnesia-zirconia. More preferably a low acidity refractory oxide binder material, which is essentially free of alumina, is used. Examples of these binder materials are silica, zirconia, titanium dioxide, germanium dioxide, boria and mixtures of two or more of these of which examples are listed above. The most preferred binder is silica.

- 12 -

A preferred class of dewaxing catalysts comprise intermediate zeolite crystallites as described above and a low acidity refractory oxide binder material which is essentially free of alumina as described above, wherein the surface of the aluminosilicate zeolite crystallites has been modified by subjecting the aluminosilicate zeolite crystallites to a surface dealumination treatment. A preferred dealumination treatment is by contacting an extrudate of the binder and the zeolite with an aqueous solution of a fluorosilicate salt as described in for example US-A-5157191 or WO-A-0029511. Examples of suitable dewaxing catalysts as described above are silica bound and dealuminated Pt/ZSM-5, and more preferably silica bound and dealuminated Pt/ZSM-23, silica bound and dealuminated Pt/ZSM-12, silica bound and dealuminated Pt/ZSM-22, as for example described in WO-A-0029511 and EP-B-832171.

Catalytic dewaxing conditions are known in the art and typically involve operating temperatures in the range of from 200 to 500 °C, suitably from 250 to 400 °C, hydrogen pressures in the range of from 10 to 200 bar, preferably from 40 to 70 bar, weight hourly space velocities (WHSV) in the range of from 0.1 to 10 kg of oil per litre of catalyst per hour (kg/l/hr), suitably from 0.2 to 5 kg/l/hr, more suitably from 0.5 to 3 kg/l/hr and hydrogen to oil ratios in the range of from 100 to 2,000 litres of hydrogen per litre of oil. By varying the temperature between 275, suitably between 315 and 375 °C at between 40-70 bars, in the catalytic dewaxing step it is possible to prepare base oils having different pour point specifications varying from suitably -10 to -60 °C.

The effluent or separate boiling fractions of the catalytic or solvent dewaxing step are optionally subjected to an additional hydrogenation step, also

- 13 -

referred to as a hydrofinishing step for example if the effluent contains olefins or when the product is sensitive to oxygenation or when colour needs to be improved. This step is suitably carried out at a temperature between 180 and 380 °C, a total pressure of between 10 to 250 bar and preferably above 100 bar and more preferably between 120 and 250 bar. The WHSV (Weight hourly space velocity) ranges from 0.3 to 2 kg of oil per litre of catalyst per hour (kg/l.h).

The hydrogenation catalyst is suitably a supported catalyst comprising a dispersed Group VIII metal. Possible Group VIII metals are cobalt, nickel, palladium and platinum. Cobalt and nickel containing catalysts may also comprise a Group VIB metal, suitably molybdenum and tungsten. Suitable carrier or support materials are low acidity amorphous refractory oxides. Examples of suitable amorphous refractory oxides include inorganic oxides, such as alumina, silica, titania, zirconia, boria, silica-alumina, fluorided alumina, fluorided silica-alumina and mixtures of two or more of these.

Examples of suitable hydrogenation catalysts are nickel-molybdenum containing catalyst such as KF-847 and KF-8010 (AKZO Nobel) M-8-24 and M-8-25 (BASF), and C-424, DN-190, HDS-3 and HDS-4 (Criterion); nickel-tungsten containing catalysts such as NI-4342 and NI-4352 (Engelhard) and C-454 (Criterion); cobalt-molybdenum containing catalysts such as KF-330 (AKZO-Nobel), HDS-22 (Criterion) and HPC-601 (Engelhard). Preferably platinum containing and more preferably platinum and palladium containing catalysts are used. Preferred supports for these palladium and/or platinum containing catalysts are amorphous silica-alumina. Examples of suitable silica-alumina carriers are disclosed in WO-A-9410263. A preferred catalyst comprises an alloy of palladium and platinum preferably supported on an amorphous silica-

- 14 -

alumina carrier of which the commercially available catalyst C-624 of Criterion Catalyst Company (Houston, TX) is an example.

5 The dewaxed product is suitably separated into one or more base oil products having different viscosities by means of distillation, optionally in combination with an initial flashing step. The separation into the various fractions may suitably be performed in a vacuum distillation column provided with side strippers to
10 separate the fraction from said column. In this mode it is found possible to obtain for example a base oil having a viscosity between 2-3 cSt, a base oil having a viscosity between 4-6 cSt and a base oil having a viscosity between 7-10 cSt product simultaneously from a
15 single waxy raffinate product (viscosities as kinematic viscosity at 100 °C). By straightforward optimising the product slate and minimising the amount of non-base oil intermediate fractions it has been found possible to prepare base oils in a sufficiently high yield having a
20 good Noack volatility properties. For example, base oils having a kinematic viscosity at 100 °C of between 3.5 and 6 cSt have been obtained which have a Noack volatility of between 6 and 14 wt%.

It has been found that a lubricating base oil can be
25 prepared starting from this waxy raffinate product which base oil comprises preferably at least 98 wt% saturates, more preferably at least 99.5 wt% saturates and most preferably at least 99.9 wt%. This saturates fraction in the base oil comprises between 10 and 40 wt% of cyclo-
30 paraffins. Preferably the content of cyclo-paraffins is less than 30 wt% and more preferably less than 20 wt%. Preferably the content of cyclo-paraffins is at least 12 wt%. The unique and novel base oils are further characterized in that the weight ratio of 1-ring cyclo-
35 paraffins relative to cyclo-paraffins having two or more

- 15 -

rings is greater than 3 preferably greater than 5. It was found that this ratio is suitably smaller than 15.

The cyclo-paraffin content as described above is measured by the following method. Any other method
5 resulting in the same results may also be used. The base oil sample is first separated into a polar (aromatic) phase and a non-polar (saturates) phase by making use of a high performance liquid chromatography (HPLC) method IP368/01, wherein as mobile phase pentane is used instead
10 of hexane as the method states. The saturates and aromatic fractions are then analyzed using a Finnigan MAT90 mass spectrometer equipped with a Field desorption/Field Ionisation (FD/FI) interface, wherein FI (a "soft" ionisation technique) is used for the semi-
15 quantitative determination of hydrocarbon types in terms of carbon number and hydrogen deficiency. The type classification of compounds in mass spectrometry is determined by the characteristic ions formed and is normally classified by "z number". This is given by the
20 general formula for all hydrocarbon species: C_nH_{2n+z} . Because the saturates phase is analysed separately from the aromatic phase it is possible to determine the content of the different (cyclo)-paraffins having the same stoichiometry. The results of the mass spectrometer
25 are processed using commercial software (poly 32; available from Sierra Analytics LLC, 3453 Dragoo Park Drive, Modesto, California GA95350 USA) to determine the relative proportions of each hydrocarbon type and the average molecular weight and polydispersity of the
30 saturates and aromatics fractions.

The base oil composition preferably has a content of aromatic hydrocarbon compounds of less than 1 wt%, more preferably less than 0.5 wt% and most preferably less than 0.1 wt%, a sulphur content of less than 20 ppm and a

- 16 -

nitrogen content of less than 20 ppm. The pour point of the base oil is preferably less than -30 °C and more preferably lower than -40 °C. The viscosity index is higher than 120. It has been found that the novel base oils typically have a viscosity index of below 140.

The base oils itself may find application as part of for example an Automatic Transmission Fluids (ATF), automotive (gasoline or diesel) engine oils, turbine oils, hydraulic oils, electrical oils or transformer oils and refrigerator oils.

The invention will be illustrated with the following non-limiting examples.

Example 1

A waxy raffinate product was obtained by feeding continuously a C₅-C₇₅₀ °C⁺ fraction of the Fischer-Tropsch product, as obtained in Example VII using the catalyst of Example III of WO-A-9934917 to a hydrocracking step (step (a)). The feed contained about 60 wt% C₃₀⁺ product. The ratio C₆₀⁺/C₃₀⁺ was about 0.55. In the hydrocracking step the fraction was contacted with a hydrocracking catalyst of Example 1 of EP-A-532118.

The effluent of step (a) was continuously distilled to give lights, fuels and a residue "R" boiling from 370 °C and above. The yield of gas oil fraction on fresh feed to hydrocracking step was 43 wt%. The main part of the residue "R" was recycled to step (a) and a remaining part was separated by means of a vacuum distillation into a waxy raffinate product having the properties as in Table 1 and a fraction boiling above 510 °C.

The conditions in the hydrocracking step (a) were: a fresh feed Weight Hourly Space Velocity (WHSV) of 0.8 kg/l.h, recycle feed WHSV of 0.2 kg/l.h, hydrogen gas rate = 1000 Nl/kg, total pressure = 40 bar, and a reactor temperature of 335 °C.

- 17 -

Table 1

Density at 70°C (kg/m ³)	779.2
vK@100 (cSt)	3.818
pour point (°C)	+18
Boiling point data as temperature at which a wt% is recovered.	5% 355 °C
	10% 370 °C
	50% 419 °C
	90% 492 °C
	95% 504 °C

Example 2

The waxy raffinate product of Example 1 was dewaxed to prepare a base oil by contacting the product with a dealuminated silica bound ZSM-5 catalyst comprising 0.7% by weight Pt and 30 wt% ZSM-5 as described in Example 9 of WO-A-0029511. The dewaxing conditions were 40 bar hydrogen, WHSV = 1 kg/l.h and a temperature of 340 °C.

The dewaxed oil was distilled into three base oil fractions: boiling between 378 and 424 °C (yield based on feed to dewaxing step was 14.2 wt%), between 418-455 °C (yield based on feed to dewaxing step was 16.3 wt%) and a fraction boiling above 455 °C (yield based on feed to dewaxing step was 21.6 wt%). See Table 2 for more details.

- 18 -

Table 2

	Light Grade	Medium Grade	Heavy Grade
density at 20 °C	805.8	814.6	822.4
pour point (°C)	< -63	< -51	- 45
kinematic viscosity at 40 °C (cSt)		19.06	35.0
kinematic viscosity at 100 °C (cSt)	3.16	4.144	6.347
VI	n.a.	121	134
Noack volatility (wt%)	n.a.	10.8	2.24
sulphur content (ppm)	< 1 ppm	< 1 ppm	< 5 ppm
saturates (%w)	n.a.	99.9	n.a.
Content of cyclo- paraffins (wt%)(*)	n.a.	18.5	n.a.
Dynamic viscosity as measured by CCS at -40 °C	n.a.	3900 cP	n.a.

(*) as determined by means of a Finnigan MAT90 mass spectrometer equipped with a Field desorption/field ionisation interface on the saturates fraction of said base oil.

5 n.a.: not applicable

n.d.: not determined

Example 3

Example 2 was repeated except that the dewaxed oil was distilled into the different three base oil products of which the properties are presented in Table 3.

10

- 19 -

Table 3

	Light Grade	Medium Grade	Heavy Grade
density at 20 °C	809.1	817.2	825.1
pour point (°C)	< -63	< -51	- 39
kinematic viscosity at 40 °C (cSt)		23.32	43.01
kinematic viscosity at 100 °C (cSt)	3.181	4.778	7.349
VI	n.a.	128	135
Noack volatility (wt%)	n.a.	7.7	n.a.
sulphur content (ppm)	< 5 ppm	< 5 ppm	< 5 ppm
saturates (%w)		99.0	
Dynamic viscosity as measured by CCS at -40 °C		5500 cP	
Yield based on feed to cat dewaxing step (wt%)	15.3	27.4	8.9

Example 4

Example 2 was repeated except that the that the dewaxed oil was distilled into the different three base oil products and one intermediate raffinate (I.R.) of which the properties are presented in Table 4.

- 20 -

Table 4

	Light Grade	I.R.	Medium Grade	Heavy Grade
density at 20 °C	806	811.3	817.5	824.5
pour point (°C)	< -63	-57	< -51	- 39
Kinematic viscosity at 40 °C (cSt)	10.4		23.51	42.23
Kinematic viscosity at 100 °C (cSt)	2.746	3.501	4.79	7.24
VI	103		127	135
Noack volatility	n.a.		6.8	1.14
sulphur content (ppm)	< 5 ppm		< 5 ppm	< 5 ppm
Saturates (%w)	n.d.		99.5	
Dynamic viscosity as measured by CCS at -40 °C			5500 cP	
Yield based on CDW feed	22.6	8.9	22.6	11.1

n.a.: not applicable

n.d.: not determined

Examples 2-4 illustrate that from the waxy raffinate product as obtained by the process of the present invention base oils are prepared in a high yield and wherein the base oils have excellent viscometric properties.

C L A I M S

1. Process to prepare a waxy raffinate product by
(a) hydrocracking/hydroisomerising a Fischer-Tropsch
derived feed, wherein weight ratio of compounds having at
least 60 or more carbon atoms and compounds having at
least 30 carbon atoms in the Fischer-Tropsch product is
at least 0.2 and wherein at least 30 wt% of compounds in
the Fischer-Tropsch derived feed have at least 30 carbon
atoms,

(b) isolating from the product of step (a) a waxy
raffinate product having a T10 wt% boiling point of
between 200 and 450 °C and a T90 wt% boiling point of
between 400 and 650 °C.

2. Process according to claim 1, wherein at least 50 wt%
of compounds in the Fischer-Tropsch derived feed have at
least 30 carbon atoms.

3. Process according to any one of claims 1-2, wherein
the weight ratio of compounds having at least 60 or more
carbon atoms and compounds having at least 30 carbon
atoms in the Fischer-Tropsch derived feed is at least
0.4.

4. Process according to any one of claims 1-3, wherein
the Fischer-Tropsch derived feed is derived from a
Fischer-Tropsch product comprising a C₂₀⁺ fraction having
an ASF-alpha value (Anderson-Schulz-Flory chain growth
factor) of at least 0.925.

5. Process according to any one of claims 1-4, wherein
the conversion in step (a) is between 25 and 70 wt%.

6. Process according to any one of claims 1-5, wherein
the T90wt% boiling point of the waxy raffinate product is
below 550 °C.

7. Process according to any one of claims 1-6, wherein the waxy raffinate product has a kinematic viscosity at 100 °C of between 3 and 10 cSt.
- 5 8. Use of the waxy raffinate product as obtained in a process according to any one of claims 1-7 to prepare a lubricating base oil.
9. Use according to claim 8 wherein the base oil is prepared by catalytically dewaxing the waxy raffinate product.
- 10 10. Use according to claim 9, wherein the cyclo-paraffin content in the saturates fraction of the lubricating base oil is between 12 and 20 wt%.
11. Method to transport the product as obtained in the process according to any one of claims 1-7, wherein the
15 waxy raffinate product has a pour point of above 0 °C and wherein the waxy raffinate is kept in the solid state under nitrogen blanketing during transport.

INTERNATIONAL SEARCH REPORT

International Application No.

PC1/EP 02/02449

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C10G65/12 C10G65/04

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C10G

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO 96 03359 A (MOBIL OIL CORP) 8 February 1996 (1996-02-08) claim 1 page 10, line 11 - line 13 page 10, line 18 - line 22 page 11, line 13 - line 17 table 1 example 1	1,2,5, 7-9
A	WO 97 21788 A (EXXON RESEARCH ENGINEERING CO) 19 June 1997 (1997-06-19) claims 1,2,8 page 4, paragraph 1 table 1	1,5,8
	--- -/-- ---	



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

* Special categories of cited documents:

A document defining the general state of the art which is not considered to be of particular relevance

E earlier document but published on or after the international filing date

L document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

O document referring to an oral disclosure, use, exhibition or other means

P document published prior to the international filing date but later than the priority date claimed

T later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

X document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

Y document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

G document member of the same patent family

Date of the actual completion of the international search

22 July 2002

Date of mailing of the international search report

29/07/2002

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax (+31-70) 340-3016

Authorized officer

De Herdt, O

INTERNATIONAL SEARCH REPORT

International Application No

PCT/EP 02/02449

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 6 179 994 B1 (CLARK JANET R ET AL) 30 January 2001 (2001-01-30) claims 1-3,5,11 column 6, line 34 - line 39 column 8, line 33 - line 35 -----	1,5
A	WO 00 15736 A (EXXON RESEARCH ENGINEERING CO) 23 March 2000 (2000-03-23) the whole document -----	1-3,5,7, 9
A	US 6 165 949 A (HABEEB JACOB J ET AL) 26 December 2000 (2000-12-26) the whole document -----	1-3,5,7, 9
A	EP 0 776 959 A (SHELL INT RESEARCH) 4 June 1997 (1997-06-04) cited in the application the whole document -----	1-3,5,7, 9

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/EP 02/02449

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
WO 9603359	A	08-02-1996	US 5362378 A	08-11-1994
			WO 9603359 A1	08-02-1996
			AU 695832 B2	20-08-1998
			AU 7474494 A	22-02-1996
			DE 69421100 D1	11-11-1999
			DE 69421100 T2	20-01-2000
			EP 0783470 A1	16-07-1997
			SG 66235 A1	20-07-1999
WO 9721788	A	19-06-1997	AU 1053597 A	03-07-1997
			BR 9611898 A	16-05-2000
			CA 2237068 A1	19-06-1997
			EP 0876446 A1	11-11-1998
			JP 2000502135 T	22-02-2000
			NO 982629 A	08-06-1998
			TW 442565 B	23-06-2001
			WO 9721788 A1	19-06-1997
			US 6096940 A	01-08-2000
			ZA 9609890 A	12-06-1997
US 6179994	B1	30-01-2001	AU 5693899 A	27-03-2000
			BR 9913412 A	22-05-2001
			EP 1144552 A2	17-10-2001
			NO 20011000 A	04-05-2001
			WO 0014184 A2	16-03-2000
			US 6375830 B1	23-04-2002
WO 0015736	A	23-03-2000	US 6332974 B1	25-12-2001
			AU 5680899 A	03-04-2000
			BR 9913583 A	22-05-2001
			EP 1144551 A2	17-10-2001
			NO 20011245 A	10-05-2001
			WO 0015736 A2	23-03-2000
US 6165949	A	26-12-2000	AU 5690299 A	27-03-2000
			BR 9913410 A	22-05-2001
			EP 1114132 A2	11-07-2001
			NO 20011123 A	02-05-2001
			US 2002086803 A1	04-07-2002
			WO 0014188 A2	16-03-2000
EP 0776959	A	04-06-1997	EP 0776959 A2	04-06-1997
			AU 705415 B2	20-05-1999
			AU 7198896 A	05-06-1997
			CA 2191290 A1	29-05-1997
			CN 1167811 A	17-12-1997
			JP 9221685 A	26-08-1997
			SG 74001 A1	18-07-2000
			TW 416981 B	01-01-2001
			ZA 9609906 A	28-05-1997